

X-ray Diffraction Studies of Structures of Lutetium Aluminum Oxide

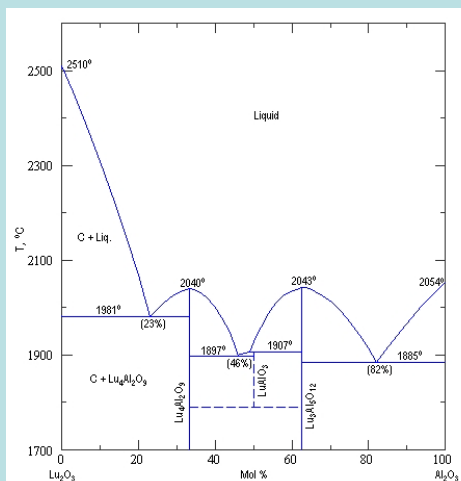
C.J. Rawn¹, S.A. Speakman¹, M.A. Spurrier², and P. Szupryczynski²

¹Metals and Ceramics Division, Oak Ridge National Laboratory, Oak Ridge, TN 37831 USA

²CTI Molecular Imaging, 810 Innovation Drive, Knoxville, TN 37923 USA

Introduction

Single crystal cerium-doped LuAlO₃ has the potential of being used in numerous applications including nuclear medicine, security monitoring, geophysical exploration, and non-destructive testing where efficient and fast scintillators are required. The calculated phase diagram for the Lu₂O₃-Al₂O₃ system [1] is shown below.



PDFC fig. 10396 system Lu₂O₃-Al₂O₃

In the above phase diagram three compounds are shown: Lu₃Al₅O₁₂ which has the garnet structure [2] (LuAG), LuAlO₃ which has the perovskite structure [3] (LuAP), and Lu₄Al₂O₉. The work here summarizes the results from five different crystal growth runs.

Experimental

Cerium-doped Lu₂AlO₃ (LuAP) was grown by the Czochralski method in inductively heated iridium crucibles in a nitrogen atmosphere to which small amounts of oxygen, ranging from 0.1 to 0.5%, was added. The starting materials were Lu₂O₃ and Al₂O₃ with a purity of at least 99.99%, with melt Ce concentrations ranging from 0.4 to 1 atomic percent. The residual melt is the fraction of the initial charge retained in the crucible after the growth run is complete.

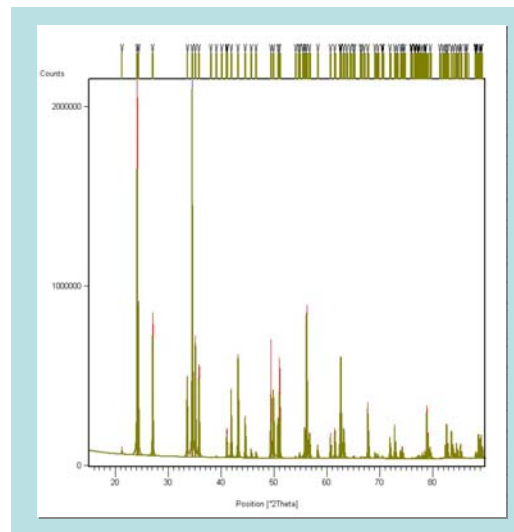
The samples were ground in an agate mortar pestle and screened down to <32 μm (450 mesh). The powder samples were contained in deep well sample holders for data collection. X-ray powder diffraction data were collected on a PANalytical MPD X'Pert PRO θ-θ diffractometer with an X'Celerator Real-Time Multiple Strip detector using CuKα radiation. Data were collected between 15 and 90° two-theta. The scan speed was either 1700 s/step or 200s/step and is given with the results in Table 1.

The data were refined using the PANalytical Rietveld package Highscore Plus.

Table 1. Data collection conditions, phase(s) present, and physical appearance

run	Rate (s/step)	Sample ID	Phase(s) present	color
1	1700	single crystal	LuAP	colorless
	200	residual	LuAP	white (powder)
2	200	single crystal	LuAG+ Lu ₄ Al ₂ O ₉	yellow
	200	residual	LuAG + Lu ₄ Al ₂ O ₉	yellow (powder)
3	1700	single crystal	LuAP	colorless
	200	residual	LuAP	white (powder)
4	1700	single crystal	LuAP	colorless
	200	residual	LuAG + Lu ₄ Al ₂ O ₉	yellow (powder)
5	200	residual	LuAG, Lu ₂ O ₃ , and Lu ₄ Al ₂ O ₉	pink (powder)

Results



Rietveld refinements on the data collected on the powdered samples from the colorless single crystals showed single phase LuAlO₃. To model the x-ray powder diffraction pattern the atomic positions for YAlO₃ [4] were used. The slow rate of data collection reduced the noise so that Lu₃Al₅O₁₂ could have been detected if present at approximately 0.2 wt% or greater. Both Lu₃Al₅O₁₂ and another phase, most like Lu₄Al₂O₉, were observed in the data collected on the samples that were yellow in color.

Attempts to model Lu₄Al₂O₉ as a secondary phase in the powder pattern using the atomic coordinates from Y₄Al₂O₉ were unsatisfactory. One crystal growth run resulted in a pink residual powder and the x-ray diffraction data revealed three phases (Lu₂O₃, Lu₄Al₂O₉, and LuAlO₃).

References

1. P.Wu and A.D. Pelton, *J. Alloys Compd.*, **179**, 259-287 (1992).
2. F. Euler and J.A. Bruce, *Acta Cryst.*, **19**,971-978 (1965).
3. P.D. Dernier and R.G. Maines, *Mat. Res. Bull.*, **6**, 433-440 (1971)
4. .

Acknowledgements

This research was sponsored by the Assistant Secretary for Energy Efficiency and Renewable Energy, Office of FreedomCAR and Vehicle Technologies, as part of the High Temperature Materials Laboratory User Program, Oak Ridge National Laboratory, managed by UT-Battelle, LLC, for the U.S. Department of Energy under contract number DE-AC05-00OR22725.